

## LETTERS TO THE EDITOR

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### Conversion of Vitamin B<sub>12b</sub> into Vitamin B<sub>12</sub>

SIR.—The spectra<sup>1,2</sup> of vitamins B<sub>12</sub> and B<sub>12b</sub>, resemble each other in general character, but differ considerably in the position and magnitude of particular features. A structural similarity is thus evident between the two compounds<sup>3</sup> which is reflected in certain of their chemical reactions. On treatment with cyanide ion<sup>4</sup> in aqueous solution, both compounds form purple cyanide complexes having the *same* absorption spectrum, which is markedly different from those of the two vitamins. On removing cyanide from the complex, vitamin B<sub>12</sub> reverts to its original state, as shown by its absorption spectrum, microbiological activity, chromatographic behaviour<sup>5</sup>, and general crystallographic appearance. On removing the cyanide from the B<sub>12</sub>-cyanide product, however, the spectrum does not revert to that of B<sub>12b</sub>, but is strikingly similar to that of B<sub>12</sub>. The regenerated material, moreover, shows further properties similar to those of B<sub>12</sub>, viz. microbiological activity against *Lactobacillus lactis* Dorner and behaviour on paper chromatography<sup>5</sup>. These observations lead to the conclusion that B<sub>12b</sub> may be converted, through intermediate formation of a cyanide complex, into a substance which, as far as can be ascertained at present, is identical with vitamin B<sub>12</sub>. As B<sub>12</sub> appears to form the major constituent of the B<sub>12</sub> group of factors in materials derived from the *in vitro* fermentation of micro-organisms, the above conversion assumes preparative importance.

A full account of this work is being submitted for publication in your Journal.

It is a pleasure to acknowledge the encouragement of the Directors of The British Drug Houses, Ltd., in this work.

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